

Property Library for Propane

LibPropan

Prof. Hans-Joachim Kretschmar
Dr. Sebastian Herrmann
Prof. Matthias Kunick
Ines Jaehne
R. Krause

Property Functions

Functional Dependence	Function Name	Call from Fortran program	Property or Function	Unit of the result
$a = f(p, t, x)$	a_ptx_PROP	APTXPROP(P,T,X)	Thermal diffusivity	m ² /s
$c_p = f(p, t, x)$	cp_ptx_PROP	CPPTXPROP(P,T,X)	Specific isobaric heat capacity	kJ/(kg K)
$\eta = f(p, t, x)$	eta_ptx_PROP	ETAPTXPROP(P,T,X)	Dynamic viscosity	Pa s
$h = f(p, t, x)$	h_ptx_PROP	HPTXPROP(P,T,X)	Specific enthalpy	kJ/kg
$\kappa = f(p, t, x)$	kappa_ptx_PROP	KAPTXPROP(P,T,X)	Isentropic exponent	-
$\lambda = f(p, t, x)$	lambda_ptx_PROP	LAMPTXPROP(P,T,X)	Thermal conductivity	W/(m K)
$\nu = f(p, t, x)$	ny_ptx_PROP	NYPTXPROP(P,T,X)	Kinematic viscosity	m ² /s
$p_{\text{mel}} = f(t)$	pmel_t_PROP	PMELPROP(T)	Melting pressure from temperature	bar
$p_s = f(t)$	ps_t_PROP	PSTPROP(T)	Vapor pressure from temperature	bar
$Pr = f(p, t, x)$	Pr_ptx_PROP	PRPTXPROP(P,T,X)	<i>Prandtl</i> -Number	-
$\rho = f(p, t, x)$	rho_ptx_PROP	ROPTXPROP(P,T,X)	Density	kg/m ³
$s = f(p, t, x)$	s_ptx_PROP	SPTXPROP(P,T,X)	Specific entropy	kJ/(kg K)
$t = f(p, h)$	t_ph_PROP	TPHPROP(P,H)	Backward function: Temperature from pressure and enthalpy	°C
$t = f(p, s)$	t_ps_PROP	TPSPROP(P,S)	Backward function: Temperature from pressure and entropy	°C
$t_{\text{mel}} = f(p)$	tmel_p_PROP	TMELPROP(P)	Melting temperature from pressure	°C
$t_s = f(p)$	ts_p_PROP	TSPPROP(P)	Saturation temperature from pressure	°C
$u = f(p, t, x)$	u_ptx_PROP	UPTXPROP(P,T,X)	Specific internal energy	kJ/kg
$v = f(p, t, x)$	v_ptx_PROP	VPTXPROP(P,T,X)	Specific volume	m ³ /kg
$w = f(p, t, x)$	w_ptx_PROP	WPTXPROP(P,T,X)	Isentropic speed of sound	m/s
$x = f(p, h)$	x_ph_PROP	XPHPROP(P,H)	Backward function: Vapor fraction from pressure and enthalpy	kg/kg
$x = f(p, s)$	x_ps_PROP	XPSPROP(P,S)	Backward function: Vapor fraction from pressure and entropy	kg/kg

Units: t in °C
 p in bar
 x in (kg of saturated steam)/(kg wet steam)

Range of validity

Temperature range: from $t_{\text{mel}}(p)$ to 326.85 °C
Pressure range: from $p_s(t) = 1.7517 \times 10^{-9}$ bar to 1000 bar

Reference state

$h = 200$ kJ/kg and $s = 1$ kJ/(kg K) at $t = 0$ °C on the boiling curve ($x = 0$)

Details on the vapor fraction x

The wet steam region is calculated automatically by the subprograms. For this purpose the following fixed details on the vapor fraction x are to be considered:

Single-phase region

If the state point to be calculated is located in the single-phase region (liquid or superheated steam) $x = -1$ must be entered as a pro-forma value.

Wet-steam region

If the state point to be calculated is located in the wet steam region, a value for x between 0 and 1 ($x = 0$ for saturated liquid, $x = 1$ for saturated steam) must be entered. In this case, the backward functions result in the appropriate value between 0 and 1 for x . When calculating wet steam either the given value for t and $p = -1000$ or the given value for p and $t = -1000$ and in both cases the value for x between 0 and 1 must be entered.

If p and t and x are entered as given values, the program considers p and t to be appropriate to represent the vapor pressure curve. If this is not the case the calculation for the property of the chosen function results in -1000 .

Wet steam region: Temperature range from $t_t = -187.625$ °C to $t_c = 96.74$ °C

Pressure range from $p_s(t_t) = 1.7517 \times 10^{-9}$ bar to $p_c = 42.512$ bar

Note.

If the calculation results in -1000 , the values entered represent a state point beyond the range of validity of LibPropan. For further information on each function and its range of validity see Chapter 3. The same information may also be accessed via the online help pages.