

Zittau/Goerlitz University of Applied Sciences Department of Technical Thermodynamics http://thermodynamics.hs-zigr.de

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EU Project AA-CAES

Work Package 4: Thermophysical Properties

Contributions to:

- Task 4.2:Development of a data base
- Task 4.3:Identification / development of suitable
models, exploitation of the results

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Collection of Data from the AA-CAES Experimental Groups

▶ Dry Air
- *p*-*p*-*T* data from PTB, ICSTM and RUB
Speed of sound data from ICSTM
Viscosity data from RUB
▶ Humid Air
- *ψ* H₂O,s data from ICSTM
- *p*-*p*-*T* data from ICSTM and PTB
Preliminary data for density from RUB
Preliminary data for viscosity from RUB





Comparisons with Experimental Data – Current State

Library LibHuAir

- Model of ideal mixture of the real fluids dry air from Lemmon et al. (2000) steam from IAPWS-IF97
- Poynting correction for saturation pressure of steam
- Dissociation from VDI-4670

Library LibHuGas

- Model of ideal mixture of the real fluids
 - N₂ from Span et al. (2000)
 - O₂ from Schmidt and Wagner (1987)
 - Ar from Tegeler et al. (1999)
 - Steam from IAPWS-95
- Poynting correction for saturation pressure of steam
- Dissociation from VDI-4670

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Algorithm of Hyland and Wexler (1983)

- Virial equation for the mixture
- Using enhancement factor for saturation pressure of steam

Program MoistAirTab

- Model of Hyland and Wexler (1983)

Algorithm of Nelson (2001)

- Modification and extension of the model of Hyland and Wexler (1983)

Program Refprop 7.0 (for dry air)

- Fundamental equation for the mixture dry air of Lemmon et al. (2000)
 - \rightarrow Refprop_pure
- Multifluid mixing model of the components N2, O2, and Ar of Lemmon et al. (2000)
 - $\rightarrow \text{Refprop}_\text{mix}$



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Mareike or Hans-Joachim

Perhaps Fred has some info. Mine is limited to the following from the 2nd edition appendix V which I previously sent.

Your curves are similar in shape to those I have plotted up to 5 MPa. Note that if you extend the plots to 173 K you will have higher peaks but this is probably well below your temperature of interest.

The following quotes J.A. Goff from page 340 of Smithsonian Meteorological Tables, 6th revised edition (1958).

"Within the temperature range -100°C to 60°C and 5000 Pa to 110,000 Pa the enhancement factor lies between 1.0000 and 1.0089. These departures from unity may be ascribed to three separate though not unrelated effects:

(a) the effect of dissolved gases on the properties of the condensed (liquid or solid H2O) phase,

(b) the effect of pressure on the properties of the condensed phase, and

(c) the effect of intermolecular force (gas imperfections) on the properties of the moist air itself. While it is true that these departures are small enough to be disregarded in rough calculations, it should be kept in mind that the error thus committed may well exceed the probable error of the saturation pressure data themselves."

I have not seen the original Goff and Gratch research reports of the 1940s. My recollection of the Hyland Wexler reports regarding the enhancement factor is that they used the Goff methodology and used updated data.

Don Gatley







