

### ZITTAU/GÖRLITZ

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# **Property Libraries**

for Water and Steam, and other Working Fluids for Calculating Heat Cycles, Turbines, Heat Pumps, and Refrigeration Processes

#### Steam, Water, and Ice

#### Library LibIF97

- Industrial Formulation IAPWS-IF97 (Revision 2007)
- Supplementary Standards IAPWS-IF97-S01 IAPWS-IF97-S03ref IAPWS-IF97-S04 IAPWS-IF97-S05
- IAPWS Revised Advisory Note No. 3 on Thermodynamic Derivatives (2008)

#### Library LibICE

- Ice from IAPWS-06
- Melting line and sublimation line from IAPWS-08
- Water from IAPWS-IF97
- Steam from IAPWS-95 and IAPWS-IF97

#### **Humid Combustion Gas Mixtures**

#### Library LibHuGas

Model: Ideal mixture of the real fluids:

CO<sub>2</sub> - Span and Wagner (1994) - Schmidt and Wagner (1995)

O<sub>2</sub> - Schmidt an H<sub>2</sub>O - IAPWS-95

- Tegeler et al. (1999)

- Span (2000)

and of the ideal Gases:

SO<sub>2</sub>, CO, Ne (Bücker et al., 2003)

Consideration of:

- Condensation of steam
- Dissociation and poynting effect

#### Library LibIDGAS

Model: Ideal gas mixture from VDI-Guideline 4670

#### Humid Air

#### Library LibHuAir

Model: Ideal mixture of the real fluids:

- Dry air from Lemmon et al. (2000)
- Steam, water, and ice from IAPWS-IF97 and IAPWS-06
  - Consideration of:
- Condensation and freezing
- Dissociation from VDI-Guideline 4670 (2003)

of steam

 Poynting effect from ASHRAE RP-1485 (2009)

#### Ideal **Gas Mixtures**

#### Library LibldGasMix

Model: Ideal mixture of the ideal gases:

Methane SO<sub>2</sub> Ne Ethane H<sub>2</sub> H<sub>2</sub>S Ethylene  $N_2$ ОН Propylene 0, CO He Propane CO n-Butane Air NΗ<sub>3</sub> Isobutane NO Benzene Methanol

> Consideration of: Dissociation from VDI-Guideline 4670 (2003)

### **Carbon Dioxide** including Dry Ice

#### Library LibCO2

Formulation of Span and Wagner (1994)

#### Seawater

#### Library LibSeaWa

IAPWS-Formulation of Feistel (2008) and IAPWS-IF97

#### Ammonia/Water -**Mixtures**

#### Library LibAmWa

IAPWS Guideline 2001 of Tillner-Roth and Friend (1998)

### **Ammonia**

#### Library LibNH3

Formulation of Tillner-Roth (1995)

#### Water/Lithium Bromide -**Mixtures**

#### Library LibWaLi

Formulation of Kim and Infante Ferreira (2004)

#### Hydrogen

#### Library LibH2

Formulation of Leachman et al. (2007)

**Propane** 

Library LibPropane

Formulation of

Lemmon et al. (2007)

### Nitrogen

#### Library LibN2

Formulation of Span et al. (2000)

#### **Siloxanes** as ORC Working Fluids

C<sub>8</sub>H<sub>24</sub>O<sub>4</sub>Si<sub>4</sub>

Octamethylcyclotetrasiloxane

### Library LibD4

 $C_{10}H_{30}O_5Si_5$ 

Decamethylcyclopentasiloxane

## Library LibD5

 $\mathbf{C_{14}H_{42}O_{5}Si_{6}}$ 

Tetradecamethylhexasiloxane

#### Library LibMD4M

C<sub>6</sub>H<sub>18</sub>OSi<sub>2</sub>

Hexamethyldisiloxane

#### Library LibMM

Formulation of Colonna et al. (2006)

 $\mathbf{C_{12}H_{36}O_6Si_6}$ 

Dodecamethylcyclohexasiloxane

### Library LibD6

C<sub>10</sub>H<sub>30</sub>O<sub>3</sub>Si<sub>4</sub>

Decamethyltetrasiloxane

#### Library LibMD2M

 $C_{12}H_{36}O_4Si_5$ 

Dodecamethylpentasiloxane

#### Library LibMD3M

 $\mathbf{C_8H_{24}O_2Si_3}$ 

Octamethyltrisiloxane

## Library LibMDM

Formulation of Colonna et al. (2008)

#### R134a

#### Library LibR134a

Formulation of

Tillner-Roth and Baehr (1994)

## Iso-Butane

### Library LibButane\_Iso

Formulation of Bücker et al. (2003)

**Liquid Coolants** 

Library LibSecRef

Liquid solutions of water with:

Ethylene glycol

Propylene glycol

Ethyl alcohol

Methyl alcohol

Potassium carbonate

Magnesium chloride

Calcium chloride

Sodium chloride

Potassium acetate

Glycerol

Formulation of the International Institute

of Refrigeration (1997)

C<sub>2</sub>H<sub>6</sub>O<sub>2</sub>

C<sub>3</sub>H<sub>8</sub>O<sub>2</sub>

C<sub>2</sub>H<sub>5</sub>OH

CH<sub>3</sub>OH

C<sub>3</sub>H<sub>8</sub>O<sub>3</sub>

K2CO3

CaCl<sub>2</sub>

MgCl<sub>2</sub>

NaCl

C<sub>2</sub>H<sub>3</sub>KO<sub>2</sub>

#### n-Butane

#### Library LibButane n

Formulation of Bücker et al. (2003)

#### Ethanol

## Library LibC2H5OH

Formulation of Schroeder et al. (2012)

#### Methanol

#### Library LibCH3OH

Formulation of de Reuck and Craven (1993)

#### Helium

#### Library LibHe

Formulation of Arp et al. (1998)

## Dry Air including Liquid Air

## Library LibRealAir

Formulation of Lemmon et al. (2000)

#### **Hydrocarbons**

C<sub>10</sub>H<sub>22</sub> Dekane

Library LibC10H22

C<sub>5</sub>H<sub>12</sub> Isopentane

### Library LibC5H12\_ISO C<sub>5</sub>H<sub>12</sub> Neopentane

Library LibC5H12\_NEO C<sub>5</sub>H<sub>14</sub> Isohexane

Library LibC5H14

C7H8 Toluene Library LibC7H8

Formulation: Lemmon and Span (2006)

#### Other Fluids

CO Carbon monoxide

Library LibCO COS Carbonyl sulfide

Library LibCOS

H<sub>2</sub>S Hydrogen sulfide

Library LibH2S

N<sub>2</sub>O Dinitrogen monoxide

Library LibN2O SO, Sulfur dioxide

Library LibSO2

C<sub>3</sub>H<sub>6</sub>O Acetone

Library LibC3H6O

Formulation: Lemmon and Span (2006)