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# Property Libraries

## for Working Fluids for Calculating Heat Cycles, Turbines, Heat Pumps, and Refrigeration Processes

### Steam, Water, and Ice

#### Library LibIF97

- Industrial Formulation IAPWS-IF97 (Revision 2007)
- Supplementary Standards IAPWS-IF97-S01 IAPWS-IF97-S03ref IAPWS-IF97-S04 IAPWS-IF97-S05
- IAPWS Revised Advisory Note No. 3 on Thermodynamic Derivatives (2008)

#### Library LibICE

- Ice from IAPWS-06
- Melting line and sublimation line from IAPWS-08
- Water from IAPWS-IF97
- Steam from IAPWS-95 and IAPWS-IF97

### Humid Combustion Gas Mixtures

#### Library LibHuGas

- Model: Ideal mixture of the real fluids:
- CO<sub>2</sub> - Span and Wagner (1994)
  - O<sub>2</sub> - Schmidt and Wagner (1995)
  - H<sub>2</sub>O - IAPWS-95
  - Ar - Tegeler et al. (1999)
  - N<sub>2</sub> - Span (2000)
- and of the ideal Gases:
- SO<sub>2</sub>, CO, Ne (Bücker et al., 2003)
- Consideration of:
- Condensation of steam
  - Dissociation and poynting effect

#### Library LibIDGAS

Model: Ideal gas mixture from VDI-Guideline 4670

### Humid Air

#### Library LibHuAir

- Model: Ideal mixture of the real fluids:
- Dry air from Lemmon et al. (2000)
  - Steam, water, and ice from IAPWS-IF97 and IAPWS-06
- Consideration of:
- Condensation and freezing of steam
  - Dissociation from VDI-Guideline 4670 (2003)
  - Poynting effect from ASHRAE RP-1485 (2009)

### Ideal Gas Mixtures

#### Library LibIdGasMix

- Model: Ideal mixture of the ideal gases:
- |                  |                  |           |
|------------------|------------------|-----------|
| Ar               | SO <sub>2</sub>  | Methane   |
| Ne               | H <sub>2</sub>   | Ethane    |
| N <sub>2</sub>   | H <sub>2</sub> S | Ethylene  |
| O <sub>2</sub>   | OH               | Propylene |
| CO               | He               | Propane   |
| CO <sub>2</sub>  | F <sub>2</sub>   | n-Butane  |
| Air              | NH <sub>3</sub>  | Isobutane |
| NO               |                  | Benzene   |
| H <sub>2</sub> O |                  | Methanol  |
- Consideration of:  
Dissociation from VDI-Guideline 4670 (2003)

### Carbon Dioxide including Dry Ice

#### Library LibCO2

Formulation of Span and Wagner (1994)

### Ammonia/Water - Mixtures

#### Library LibAmWa

IAPWS Guideline 2001 of Tillner-Roth and Friend (1998)

### Water/Lithium Bromide - Mixtures

#### Library LibWaLi

Formulation of Kim and Infante Ferreira (2004)

### Dry Air including Liquid Air

#### Library LibRealAir

Formulation of Lemmon et al. (2000)

### Seawater

#### Library LibSeaWa

IAPWS-Formulation of Feistel (2008) and IAPWS-IF97

### Ammonia

#### Library LibNH3

Formulation of Tillner-Roth (1995)

### Hydrogen

#### Library LibH2

Formulation of Leachman et al. (2007)

### Nitrogen

#### Library LibN2

Formulation of Span et al. (2000)

### Siloxanes as ORC Working Fluids

C<sub>8</sub>H<sub>24</sub>O<sub>4</sub>Si<sub>4</sub>  
Octamethylcyclotetrasiloxane

#### Library LibD4

C<sub>10</sub>H<sub>30</sub>O<sub>5</sub>Si<sub>5</sub>  
Decamethylcyclopentasiloxane

#### Library LibD5

C<sub>14</sub>H<sub>42</sub>O<sub>5</sub>Si<sub>6</sub>  
Tetradecamethylhexasiloxane

#### Library LibMD4M

C<sub>6</sub>H<sub>18</sub>O<sub>2</sub>Si<sub>2</sub>  
Hexamethyldisiloxane

#### Library LibMM

Formulation of Colonna et al. (2006)

C<sub>12</sub>H<sub>36</sub>O<sub>6</sub>Si<sub>6</sub>  
Dodecamethylcyclohexasiloxane

#### Library LibD6

C<sub>10</sub>H<sub>30</sub>O<sub>3</sub>Si<sub>4</sub>  
Decamethyltetrasiloxane

#### Library LibMD2M

C<sub>12</sub>H<sub>36</sub>O<sub>4</sub>Si<sub>5</sub>  
Dodecamethylpentasiloxane

#### Library LibMD3M

C<sub>8</sub>H<sub>24</sub>O<sub>2</sub>Si<sub>3</sub>  
Octamethyltrisiloxane

#### Library LibMDM

Formulation of Colonna et al. (2008)

### R134a

#### Library LibR134a

Formulation of Tillner-Roth and Baehr (1994)

### Propane

#### Library LibPropane

Formulation of Lemmon et al. (2007)

### Iso-Butane

#### Library LibButane\_Iso

Formulation of Bücker et al. (2003)

### n-Butane

#### Library LibButane\_n

Formulation of Bücker et al. (2003)

### Hydrocarbons

C<sub>10</sub>H<sub>22</sub> Dekane  
Library LibC10H22

C<sub>5</sub>H<sub>12</sub> Isopentane  
Library LibC5H12\_ISO

C<sub>5</sub>H<sub>12</sub> Neopentane  
Library LibC5H12\_NEO

C<sub>6</sub>H<sub>14</sub> Isohexane  
Library LibC6H14

C<sub>7</sub>H<sub>8</sub> Toluene  
Library LibC7H8

Formulation: Lemmon and Span (2006)

### Liquid Coolants

#### Library LibSecRef

- Liquid solutions of water with:
- C<sub>2</sub>H<sub>6</sub>O<sub>2</sub> Ethylene glycol
  - C<sub>3</sub>H<sub>8</sub>O<sub>2</sub> Propylene glycol
  - C<sub>2</sub>H<sub>5</sub>OH Ethyl alcohol
  - CH<sub>3</sub>OH Methyl alcohol
  - C<sub>3</sub>H<sub>8</sub>O<sub>3</sub> Glycerol
  - K<sub>2</sub>CO<sub>3</sub> Potassium carbonate
  - CaCl<sub>2</sub> Calcium chloride
  - MgCl<sub>2</sub> Magnesium chloride
  - NaCl Sodium chloride
  - C<sub>2</sub>H<sub>3</sub>KO<sub>2</sub> Potassium acetate
- Formulation of the International Institute of Refrigeration (1997)

### Ethanol

#### Library LibC2H5OH

Formulation of Schroeder et al. (2012)

### Methanol

#### Library LibCH3OH

Formulation of de Reuck and Craven (1993)

### Helium

#### Library LibHe

Formulation of Arp et al. (1998)

### Other Fluids

CO Carbon monoxide

#### Library LibCO

COS Carbonyl sulfide

#### Library LibCOS

H<sub>2</sub>S Hydrogen sulfide

#### Library LibH2S

N<sub>2</sub>O Dinitrogen monoxide

#### Library LibN2O

SO<sub>2</sub> Sulfur dioxide

#### Library LibSO2

C<sub>3</sub>H<sub>6</sub>O Acetone

#### Library LibC3H6O

Formulation: Lemmon and Span (2006)